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following molecules: 1. Br₂ 2. Methane 3. H₂S 4. Nitrogen trichloride 5. Phosphorus pentabromide 6. Boron trihydride 7. Carbon disulfide

Lewis Structure Practice #2 Draw the Lewis structure for each of the following molecules 1. Arsenic triiodide 2.

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Vsepr And Imf Homework The valence

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shell electron pair repulsion (VSEPR) theory (or "VESPER" for short) is how the geometry of a molecule is determined around a central atom.

The molecular geometry main shapes are tetrahedral, trigonal planar, trigonal pyramidal, bent, and linear and are named by measuring

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