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Substituting and multiplying by the factor 109 for the density unity kg/km³, the mass of the atmosphere is determined to be $m = 5.092 \times 10^{18}$ kg Discussion Performing the analysis with excel would yield exactly the same results. EES Solution for final result: $a = 1.2025166$ $b = -0.10167$ $c = 0.0022375$ $r = 6377$ $h = 25$ $m = 4 \pi (a^2 r^2 h + r(2a + b r) h^2 / 2 + (a + 2b r + c r^2) h^3 / 3 + (b + 2c r) h^4 / 4 + c h^5 / 5) * 1E+9$ 1-7 Pressure, Manometer, and Barometer 1-34C The pressure relative to the atmospheric pressure is called ...

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The results are: $(z) = a + bz + cz^2 = 1.20252 - 0.101674z + 0.0022375z^2$ for the unit of kg/m³ , (or, $(z) = (1.20252 - 0.101674z + 0.0022375z^2) / 109$ for the unit of kg/km³) where z is the vertical distance from the earth surface at sea level. At $z = 7$ km, the equation would give $= 0.60$ kg/m³.

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Boles specializes in heat transfer and has been involved in the analytical and numerical solution of phase change and drying of porous media. He is a member of the American Society of Mechanical Engineers, the American Society for Engineering Education, and Sigma Xi. Dr. Boles received the ASEE Meriam/Wiley Distinguished Author Award in 1992 for excellence in authorship.

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