

Chemistry Matter Change Answers Chapter 12

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1 Chapter 2.1, 2.2 Review Packet [Answer Key 2.1 The Nature of Matter Lesson Summary](#) Atoms The atom is the basic unit of matter, made up of three subatomic particles. Protons have a positive charge and neutrons carry no charge. Strong forces bind protons and neutrons together in the nucleus.

Chemistry Chapter 2 Matter And Change Worksheet Answers

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Chemistry Matter And Change Chapter 8 [!](#) The partial pressure of a gas depends on the number of moles, size of the container, and temperature and is independent of the type of gas. [!](#) At a given temperature and pressure, the partial pressure of 1mol of any gas is the same.

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Type of mixture NOT uniform throughout(clay and water mixture-clay gathers at bottom) Pure substance. Has fixed composition and differs from a mixture in that: 1- every sample has exact same characteristic physical and chemical properties used to identify substance. 2- every sample of a pure substance has exact same composition with the same makeup (i.e.: pure water is always 11.2% H and 88.8% oxygen by mass.

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TEACHER GUIDE AND ANSWERS Chemistry: Matter and Change Teacher Guide and Answers 8 9. The first ionization energies generally increase as you move left-to-right across a period. The increased nuclear charge of each successive element produces an increased hold on the valence electrons. 10. The first ionization energies generally decrease as

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